

## COMPOSITIONS FOR REMOVING HUMAN CERUMEN

This application is a continuation of PCT/US02/19756 filed June 21, 2002 entitled  
5 “Compositions for Removing Human Cerumen,” which claims priority from U.S.  
Provisional Application No. 60/302,959, filed July 3, 2001.

### Field of the Invention

The present invention generally pertains to the removal of human cerumen. More  
10 particularly, but not by way of limitation, the present invention is directed to  
advantageous compositions for assisting in the removal of human cerumen.

### Description of the Related Art

Cerumen, or ear wax, is a mixture of secretions from the ceruminous and  
15 pilosebaceous glands as well as squamae of epithelium, dust, and other debris. Cerumen  
forms a protective layer on the skin of the external ear canal. The consistency of, and thus  
the difficulty in removing, cerumen varies from individual to individual and is at least  
partially genetically determined.

Cerumen build-up and impaction in the external ear canal is a significant problem,  
20 especially for the infant and geriatric populations of the world. In the United States about  
8 million cerumen removals take place each year, and in the United Kingdom the number  
is 2 million. Individuals possessing hairy ear canals, narrow ear canals, or osteoma are  
more disposed to such build-up or impaction. In addition, some literature suggests that  
the use of cotton buds to clean the external ear canal interferes with the body's normal  
25 shedding of ear wax and epithelium and increases the chance of such build-up and

impaction. Build-up and/or impaction of ear wax may cause irritation, itching, pain, infection, or conductive hearing loss. Wax removal is necessary to alleviate these conditions. Cerumen removal is also required when it is necessary to examine the tympanic membrane.

5           Various compositions for softening or removing human cerumen are known. Several products contain carbamide peroxide (6.5%) in an anhydrous glycerin vehicle as defined in the FDA monograph part 344. These include products such as Murine® Ear Drops available from Abbott Laboratories of Columbus, Ohio; Debrox® Drops Earwax Removal Aid available from SmithKline Beecham of Pittsburgh, Pennsylvania; Bausch & 10   Lomb Earwax Remover available from Bausch & Lomb of Rochester, New York; and Flents® Earwax Remover available from Flents Products Co. of Yonkers, New York. Another commercially available product is Cerumenex® Eardrops, which is a prescription product containing triethanolamine polypeptide oleate-condensate (10%) available from the Purdue Frederick Company of Norwalk, Connecticut. Cerumenex® sometimes 15   results in irritation of the ear canal.

          In addition, the literature speaks of certain other agents that are known to be somewhat effective in softening ear wax. Such agents include glycerin (glycerol), olive oil, almond oil, mineral oil, sodium carbonate, sodium bicarbonate, hydrogen peroxide, docusate sodium, and dichlorobenzene. After softening with one of these agents, 20   irrigation with body temperature water or saline is often performed to remove the softened cerumen. Dichlorobenzene sometimes results in irritation of the ear canal.

          Compositions that facilitate the removal of ear wax have also been the subject of several patents. For example, U.S. Patent No. 3,422,186 (Sasmor) discloses cerumenolytic compositions comprising ethylene oxide-polyoxypropylene glycol 25   condensates; U.S. Patent No. 4,895,875 (Winston) discloses stabilized peroxide solutions

comprising urea peroxide and glycerin and methods of preparation and use useful for cerumen removal; U.S. Patent No. 5,296,472 (Sanchez et al.) discloses compositions comprising cyclodextrins and methods of use for cerumen removal; U.S. Patent No. 5,380,711 (Sanchez et al.) discloses oil-free “empty” cyclodextrin compositions and methods of use for cerumen removal; and U.S. Patent No. 5,480,658 (Melman) discloses compositions comprising acetic acid and boric acid in a water base useful for cleaning the external ear canal of pets.

An aqueous solution of five percent sodium bicarbonate is often made by physicians and used to effectively treat impacted cerumen. This solution may be prepared with or without glycerin. However, these solutions are not stable, and for this reason, sodium bicarbonate solutions have never been developed into commercial products. If they are used, physicians are inconvenienced by preparing the solution for each individual patient.

Despite the above-described cerumenolytic agents and compositions, a need exists for an improved composition for assisting in the removal of human cerumen that is commercially viable and that does not suffer from the limitations of currently available cerumenolytics. The present invention provides advantageous compositions to meet this need.

## Summary of the Invention

One aspect of the present invention is a composition for assisting in the removal of human cerumen that includes bicarbonate and exhibits superior stability over known sodium bicarbonate containing aqueous cerumenolytic compositions. Another aspect of the present invention is a composition for assisting in the removal of human cerumen that includes a cerumenolytically acceptable enzyme. A further aspect of the present invention

is a composition for assisting in the removal of human cerumen that includes bicarbonate and a cerumenolytically acceptable enzyme. Each of these compositions preferably also include an aqueous otologically acceptable vehicle that acts as a carrier. The compositions of the present invention are commercially viable and provide a safe and effective means of removing human cerumen from the external ear canal.

#### Brief Description of the Drawings

For a more complete understanding of the present invention, and for further objects and advantages thereof, reference is made to the following description taken in conjunction with the accompanying drawings in which:

FIG. 1 is an exploded view of a preferred device suitable for the delivery of certain preferred compositions of the present invention;

FIG. 2 is a cross-sectional view of FIG. 1;

FIG. 3 is a front view of the device of FIG. 1 in the assembled state in which the first and second parts of the composition are not mixed;

FIG. 4 is a cross-sectional view of FIG. 3;

FIG. 5 is an enlarged, cross-sectional view of the upper container of the device of FIG. 1;

FIG. 6 is a sectional view of FIG. 5 taken along line 6-6;

FIG. 7 is a top view of the safety ring of the device of FIG 1;

FIG. 8 is an exploded, cross-sectional, fragmentary view of the tubular sleeve, upper container, and cap of the device of FIG. 1;

FIG. 9 is a front view of the device of FIG. 1 in the assembled state in which the first and second parts of the composition have been mixed, the cap is removed, and the device is ready to dispense the mixed composition; and

FIG. 10 is a cross-sectional, fragmentary view of FIG. 9.

### Detailed Description of the Invention

The preferred embodiments of the present invention and their advantages are best understood by referring to FIGS. 1-10 of the drawings, like numerals being used for like and corresponding parts of the various drawings.

Unless otherwise indicated, all ingredient concentrations listed as a percentage are in units of weight/volume percent.

As used herein, the term "bicarbonate" refers to any soluble bicarbonate salt. These salts are most frequently formed with group I metals. Sodium bicarbonate, potassium bicarbonate, or mixtures thereof are the preferred bicarbonates.

As used herein, the term "otologically acceptable vehicle" refers to any substance or combination of substances that act as a carrier for an active agent or agents and that are suitable for administration to the external ear canal. By way of example, an otologically acceptable vehicle may comprise any combination of preservatives, surfactants, viscosity enhancers, penetration enhancers, buffers, sodium chloride or other salts, solubilizers, stabilizers, pH adjusters, tonicity agents, fillers, demulcents, and water. The otologically acceptable vehicle for the compositions of the present invention is preferably an aqueous vehicle.

The preferred preservatives for the compositions of the present invention include, but are not limited to, poly[dimethylimino-2-butene-1,4-diyl]chloride-alpha-[4-tris(2-hydroxyethyl)ammonium]dichloride, which is available from Onyx Chemical Corporation as Polyquarternium 1 or Onamer M<sup>TM</sup>, or from Alcon Laboratories, Inc. as Polyquad®; benzalkonium halides such as benzalkonium chloride; alexidine salts; chlorhexidine salts; hexamethylene biguanimides and their polymers; and mixtures thereof.

The surfactant of the present invention preferably is nonionic, and may include, but is not limited to, polysorbates, such as polysorbate 20 available from ICI Americas Inc. of Wilmington, Delaware under the trademark Tween® 20; 4-(1, 1, 3, 3-tetramethylbutyl) phenol/poly(oxyethylene) polymers, such as the polymer sold under the trademark Tyloxapol; poly(oxyethylene)-poly(oxypropylene) block copolymers; polyethylene glycol esters of fatty acids, such as coconut, polysorbate, polyoxyethylene, and polyoxypropylene ethers of higher alkanes (C<sub>12</sub>-C<sub>18</sub>). Examples of the preferred class include polysorbate 20 polyoxyethylene (23) lauryl ether (Brij® 35), polyoxyethylene (40) stearate (Myrj® 52), polyoxyethylene (25) propylene glycol stearate (Atlas® G2612). Brij® 35, Myrj® 52 and Atlas® G 2612 are trademarks of, and are commercially available from, ICI Americas Inc. Most preferably the nonionic surfactant is selected from poly(oxyethylene)-poly(oxypropylene) block copolymers and mixtures thereof. Such surfactant components can be obtained commercially from the BASF Corporation under the trademarks Pluronic® and Tetronic®. Such block copolymers can be generally described as polyoxyethylene/polyoxypropylene condensation polymers terminated in primary hydroxyl groups. They may be synthesized by first creating a hydrophobe of desired molecular weight by the controlled addition of propylene oxide to the two hydroxyl groups of propylene glycol. In the second step of the synthesis, ethylene oxide is added to sandwich this hydrophobe between hydrophile groups.

Preferred buffers of the present invention include, but are not limited to, citrate, phosphate, borate, acetate, Tris, and their salts and mixtures thereof. Bicarbonates may also act as a buffer for the compositions of the present invention. Other conventional buffer components may also be employed to maintain or adjust the pH of the composition.

Certain of the compositions of the present invention employ an enzyme, as will be described in greater detail hereinbelow. The enzyme may comprise a solid composition,

such as a powder or tablet, or a liquid composition. The liquid enzyme compositions of the present invention preferably comprise an enzyme, stabilizing agents, and water.

Enzyme stabilizing agents preferred for use with the compositions of the present invention include, but are not limited to, monomeric polyols, polymeric polyols, calcium ions, and borate/boric acid compound.

As used herein, the term "monomeric polyol" refers to a compound with 2 to 6 carbon atoms and at least two hydroxy groups. Examples of monomeric polyols are glycerin, propylene glycol, ethylene glycol, sorbitol and mannitol. Preferably, the monomeric polyols are selected from polyols having 2-3 carbons and at least two hydroxy groups ("2-3 carbon polyol"). Examples of 2-3 carbon polyols are glycerin, 1,2-propane diol ("propylene glycol"), 1,3-propane diol and ethylene glycol. Glycerin and propylene glycol are the most preferred 2-3 carbon polyols.

As used herein, the term "polymeric polyol" refers to a polyalkoxylated glycol with a molecular weight ranging from about 200-600 Daltons. Examples of polymeric polyols are polyethylene glycol 200 (denoting a molecular weight of 200 Daltons, "PEG 200") and PEG 400. The PEGs may optionally be monoalkoxylated. Examples of monoalkoxylated PEGs are monomethoxy PEG 200 and ethoxy PEG 400. Though these alkoxylated PEGs are not technically polyols, they are similar in structure to the non-alkoxylated PEGs; therefore, for defining purposes, they are included in the term "polymeric polyol."

If a combination of monomeric and polymeric polyols is used, the amounts of monomeric polyol and polymeric polyol will vary depending on the particular combination of polyols used. In general, such liquid enzyme compositions will require about 40 to 85% weight/volume ("% w/v") of a polyol mixture to achieve the necessary criteria for efficacious and commercially viable liquid enzyme compositions, as described

above. The ratio of monomeric to polymeric polyols is also important. In general, the monomeric polyol:polymeric polyol ratio will be from 1:5 to 5:1, with a preferred ratio being 2:1 to 1:2, weight:weight. While any of the polyols can be components of the compositions of the present invention, particular polyols may be used depending on the particular intended use. For example, propylene glycol, which has preservative activity, is a preferred monomeric polyol when the need for an additional preservative present in a liquid enzyme composition of the present invention is desired. The most preferred combination of polyols used in the compositions of the present invention are glycerin and PEG-400. The most preferred amount of the glycerin/PEG-400 combination is 25% w/v glycerin with 50% w/v PEG-400.

The liquid enzyme compositions of the present invention may also contain an effective amount of calcium ion. The calcium ion contained in the compositions of the present invention may be obtained by the addition of various calcium salts. For example, the calcium ion source may be obtained from calcium chloride, calcium acetate and calcium ascorbate or other water soluble salts of calcium. The most preferred calcium ion source is calcium chloride. As used herein, "effective amount of calcium ion" refers to that amount of calcium ion which enhances the proteolytic stability of an enzyme in the liquid enzyme compositions of the present invention. While that amount will vary depending upon the various components present, typical calcium ion concentrations will be about 1 to 90 millimolar. Preferred concentrations will be about 4.5 to 45 millimolar, and most preferred concentrations will be of from 10 to 25 millimolar.

The liquid enzyme compositions of the present invention may also contain an effective amount of a borate/boric acid compound. As used herein, "borate/boric acid compound" refers to an inorganic compound comprising boron and one or more oxygen groups, and which is either in acid or base form when dissolved in a composition of the



present invention. Sources of borate/boric acid compounds include alkali metal salts of borate, boric acid and borax. As used herein, "effective amount of a borate/boric acid compound" refers to that amount of a borate/boric acid compound contained in a liquid enzyme composition of the present invention which enhances the proteolytic stability of the enzyme. While such an amount will vary depending on other components present, the amount will be about 0.3 to 8.0% (w/v). Preferred amounts will be of from 0.5 to 2.0% (w/v). The borate/boric acid compound may also contribute to the anti-microbial preservation of the liquid enzyme compositions of the present invention to a level effective for multi-use dispensing. The solubility of the borate/boric acid compound may be limited in water. The solubility of these compounds, however, may be increased by increasing the amount of polyol employed.

Preferred demulcents for the compositions of the present invention include, but are not limited to, povidone, polyvinyl alcohol, glycerin, propylene glycol, polyethylene glycol, and cellulose derivatives such as hydroxypropyl methyl cellulose (HPMC).

A first preferred composition of the present invention includes bicarbonate and an aqueous otologically acceptable vehicle. Sodium bicarbonate is the preferred bicarbonate. Sodium bicarbonate is preferably present in the amount of about 0.5 % to about 15 %. The amount of sodium bicarbonate is most preferably about 5%. The aqueous otologically acceptable preferably includes a demulcent, a surfactant, a preservative, and a buffer, or combinations of any of the foregoing. Glycerin is a preferred demulcent, and it is preferably present in the amount of about 1 % to about 20 %. Sodium citrate.2H<sub>2</sub>O is a preferred buffer, and it is preferably present in the amount of about 0.1 % to about 8 %. Tetronic® 1304 is a preferred surfactant, and it is preferably present in the amount of about 0.05 % to about 1 %. Benzalkonium chloride is a preferred preservative, and it is preferably present in the amount of about 0.001 % to about 0.1 %. Buffers are preferably

present in a quantity sufficient to adjust the pH of the composition from about 7.5 to about 9.0.

Examples 1 and 2 hereinbelow provide in vitro data showing that the first preferred composition is efficacious in assisting in the removal of human and artificial cerumen. The first preferred composition is preferably packaged in a bottle having a syringe or dropper to dispense the composition, or in a plastic bottle that may be squeezed to dispense the composition.

To administer the composition, a user first tilts his or her head toward one shoulder or assumes a reclined position on his or her side. The user then substantially fills the external ear canal with the composition. Approximately 1 to 2 ml of the composition typically fills the external ear canal. The user keeps his or her head titled, or remains in such a reclined position, for a sufficient time period for the composition to bathe the external ear canal and to dislodge, break-up, and/or digest cerumen in the ear canal. This time period is preferably from about fifteen minutes to about thirty minutes, although in certain cases the time period may be shorter or longer. Unlike many known cerumenolytic agents, it is believed that the aqueous compositions of the present invention may not require irrigation with water or saline to complete cerumen removal due to its efficacy in dislodging, breaking-up, and digesting cerumen. Alternatively, following the termination of the bathing period, irrigation with body temperature water, saline, or other rinsing fluid may be performed to facilitate removal of cerumen from the ear canal.

A second preferred composition of the present invention includes an enzyme and an otologically acceptable vehicle. The otologically acceptable vehicle is preferably identical, or substantially similar, to the otologically acceptable vehicle described above in connection with the first preferred composition. However, the otologically acceptable

vehicle of the second preferred composition may also include an enzyme stabilizing agent. Glycerin is a preferred enzyme stabilizing agent.

The enzymes that may be utilized in the compositions and methods of the present invention include all enzymes that are useful in softening, dislodging, breaking-up, and/or  
5 digesting human cerumen in the external ear canal and cause, at most, only minor irritation to the external ear canal. For purposes of this specification, enzymes that satisfy the foregoing requirements are referred to as being "cerumenolytically acceptable". It has been discovered that preferred cerumenolytically acceptable enzymes include lipases, proteases, and amylases. In addition, combinations of lipases and proteases, lipases and  
10 amylases, and proteases and amylases may be used. The most preferred enzymes are proteases, which are also referred to herein as proteolytic enzymes.

Examples of cerumenolytically acceptable proteolytic enzymes which may be utilized in the present invention include but are not limited to pancreatin, trypsin, subtilisin, collagenase, keratinase, carboxypeptidase, papain, bromelain, aminopeptidase,  
15 elastase, Aspergillo peptidase, pronase E (from *S. griseus*), dispase (from *Bacillus polymyxa*) and mixtures thereof. If papain is used, a reducing agent, such as N-acetylcysteine, may be required. Microbially derived enzymes, such as those derived from *Bacillus*, *Streptomyces*, and *Aspergillus* microorganisms, represent another cerumenolytically acceptable preferred type of enzyme which may be utilized in the  
20 present invention. Of this sub-group of enzymes, the most preferred are the *Bacillus* derived alkaline proteases generically called "subtilisin" enzymes.

The identification, separation and purification of enzymes is known in the art. Many identification and isolation techniques exist in the general scientific literature for the isolation of enzymes, including those enzymes having proteolytic and mixed  
25 proteolytic/lipolytic/amylolytic activity. The enzymes contemplated by this invention can

be readily obtained by known techniques from plant, animal or microbial sources. In addition, with the advent of recombinant DNA techniques, it is anticipated that new sources and types of stable proteolytic enzymes will become available. Such enzymes should be considered to fall within the scope of this invention so long as they meet the criteria set forth herein.

Pancreatin, subtilisin and trypsin are preferred enzymes for use in the present invention. Pancreatin is extracted from mammalian pancreas, and is commercially available from various sources, including Scientific Protein Laboratories of Waunakee, Wisconsin; Novo Nordisk of Denmark; Sigma Chemical Co. of St. Louis, Missouri; and Boehringer Mannheim of Indianapolis, Indiana. Pancreatin USP is a mixture of proteases, lipases and amylases, and is defined by the United States Pharmacopeia ("USP"). The most preferred form of pancreatin is Pancreatin 9X. As utilized herein, the term "Pancreatin 9X" means a filtered (0.2 microns) pancreatin containing nine times the USP protease unit content. Subtilisin is derived from Bacillus bacteria and is commercially available from various commercial sources including Novo Industries; Fluka Biochemika of Buchs, Switzerland; and Boehringer Mannheim. Trypsin is a 23,800 dalton protease with 6 disulfide bridges. Trypsin can be synthesized or obtained from various sources, such as porcine, bovine or swine pancreatin. Trypsin is also available from commercial sources such as Sigma Chemical Co., Biofac Co. of the United Kingdom; and Novo Nordisk. Trypsin may vary from species to species, but in general will be highly homologous with porcine or human trypsin.

The most preferred enzymes of the present invention are the alkyl trypsins ("Al-trypsin(s)"). Al-trypsins are more stable in the liquid compositions than the native trypsin, or other native enzymes. As used herein, "Al-trypsin" refers to a covalently modified trypsin wherein one or more of its lysine epsilon-amino groups has been mono-

alkylated or di-alkylated to form the corresponding monoalkylamino or dialkylamino group. The alkyl group attached to the amine may be a primary or branched C<sub>1-12</sub> group. Preferred Al-trypsins of the present invention are those wherein the alkyl group is a primary or branched C<sub>1-4</sub> group. Alkylation of trypsin is generally performed by reductive alkylation. The degree of alkylation of the lysine epsilon-amino groups will depend on the reaction conditions of the reductive alkylation process. For example, if the reaction cycle is repeated a number of times and/or a higher reagent to enzyme ratio is used, then full alkylation, i.e., alkylation of all of the lysine epsilon-amino groups, will tend to be achieved. Al-trypsins of the present invention will preferably be greater than about 80 percent dialkylated at all of their lysine epsilon-amino groups.

The most preferred Al-trypsin is methyl trypsin ("Me-trypsin"). The most preferred Me-trypsin of the present invention will be derived from porcine tissue sources and will be greater than about 80 percent dimethylated, as described above. Methyl trypsin is preferably present in the amount of about 50 AU/ml to about 500 AU/ml.

Methyl trypsin is most preferably present in the amount of 200 AU/ml. For purposes of this specification, an "activity unit" or "AU" is defined as the amount of enzyme activity necessary to generate one microgram (mcg) of tyrosine per minute ("mcg Tyr/min"), as determined by the casein-digestion, colorimetric assay described below. 200 AU/ml is equivalent to about 600 USP units of trypsin/ml, and corresponds to about 0.25 mg of methyl trypsin per ml.

#### **Casein-digestion assay**

A 5.0 mL portion of casein substrate (0.65% casein w/v) is equilibrated for 10 minutes (min)+/- 0.5 seconds (sec) at 37° C. A 1.0 mL portion of enzyme solution (0.2 mg/ml) is then added to the casein substrate and the mixture vortexed, then incubated for 10 min+/- 0.5 sec at 37°C. After incubation, 5.0 mL of 14% trichloroacetic acid is added

and the resultant mixture immediately vortexed. The mixture is incubated for at least another 30 min, then vortexed and centrifuged for 15-20 min (approx. 2000 rpm). The supernatant of the centrifuged sample is filtered into a serum filter sampler and a 2.0 mL aliquot removed. To the 2.0 mL sample is added 5.0 mL of 5.3% sodium carbonate solution. The sample is vortexed, 1.0 mL of 0.67 N Folin's Phenol reagent is added, and the sample is immediately vortexed again, then incubated for 60 min at 37°C. The sample is then read on a visible light spectrophotometer at 660 nanometers versus purified water as the reference. The sample concentration is then determined by comparison to a tyrosine standard curve.

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Al-trypsins may be synthesized by the process of reductive alkylation of trypsin, as is more fully described in U.S. Patent No. 6,228,323, which is incorporated herein in its entirety by this reference. Me-trypsin is also available from commercial sources such as Sigma Chemical Co. and Promega Corp. (Madison, Wis.).

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During storage, some of the activity of the enzyme may be lost, depending on length of storage and temperature conditions. Thus, the enzymes of the compositions of the present invention may be prepared with initial amounts of enzyme that exceed the concentration ranges described herein. The preferred compositions of the present invention will generally contain one or more enzymes in an amount of about 50-1000 AU/ml in the combined solution. The compositions will most preferably contain about 100-500 AU/ml in the combined solution, which corresponds to pancreatin in the range of about 0.1 to 2% w/v; subtilisin in a range of about 0.01 to 0.2% w/v; trypsin in the range of about 0.01 to 0.2% w/v; and methyl trypsin in the range of about 0.01 to 0.2% w/v.

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Examples 1 and 2 hereinbelow provide in vitro data showing the second preferred composition is efficacious in assisting in the removal of human and artificial cerumen.

Because many cerumenolytically acceptable enzymes are unstable in highly aqueous solutions after a period of days or weeks, the second preferred composition is preferably prepared and packaged as two separate parts that are mixed prior to administration to the external ear canal. The first part preferably comprises the cerumenolytically acceptable enzyme and the enzyme stabilizer of the otologically acceptable vehicle, and the second part preferably comprises the remainder of the otologically acceptable vehicle. Example 3 hereinbelow describes the first part and the second part of an exemplary second preferred composition of the present invention, as well as a preferred method of making said first and second parts.

Once made, the first and second parts may be removed from their separate packages and mixed. A bottle with a syringe or dropper, or a squeezable plastic dispenser bottle, is preferably used to deliver the mixed composition to the ear as described hereinabove in connection with the first preferred composition.

Alternatively, the second preferred composition may be packaged in a single device having separate containers or compartments for the first part and the second part. Such a device preferably enables the mixing of the two parts prior to administration, and also enables the administration of the mixed composition to the external ear canal as described hereinabove in connection with the first preferred composition. FIGS. 1-10 illustrate such a device 10 suitable for the packaging of the first part 22 and the second part 26 of the second preferred composition according to a preferred embodiment of the present invention.

As shown best in the exploded views of FIGS. 1-2, device 10 generally includes a lower container 12, and upper container 14, a tubular sleeve or member 16, a cap 18, and

a safety seal 19. As shown best in FIGS. 3 and 4, in the assembled state of device 10, upper container 14 is coupled to lower container 12, tubular sleeve 16 is disposed within and coupled to upper container 14, cap 18 covers tubular sleeve 16 and is removably coupled to upper container 14, and safety seal 19 is disposed around upper container 14 below cap 18. Cap 18 and safety seal 19 are shown in dashed lines in FIG. 4 for clarity of illustration. The various portions of device 10 are preferably formed from conventional polymeric materials. Most preferably, lower container 12 is formed of low density polyethylene, upper container 14 is formed of high density polyethylene, tubular sleeve 16 is formed of Zylar, a copolymer available from NOVA Chemicals of Leominster, Massachusetts, and cap 18 is formed of polypropylene. Upper container 14 has a reservoir 20 for holding the first part 22. Lower container 12 has a reservoir 24 for holding the second part 26.

Lower container 12 includes a hollow neck 28 that includes two ring shaped edges 30 and 32. Stria 34 are located on edges 30 and 32. A shoulder 36 is located at the junction of neck 28 and reservoir 24. Sealing rings 37 are located on the internal surface of neck 28 above shoulder 36.

As shown best in FIGS. 2 and 4, upper container 14 includes a surface 82 that mates with sealing rings 37 of lower container 12 to prevent leakage of second part 26 from reservoir 24 or entry of air into reservoir 24. Upper container 14 also includes a flap 38 that surrounds neck 28 of lower container 12. As shown best in FIG. 5, flap 38 includes two ring shaped ribs 40 and 42 disposed on the internal surface 44 of flap 38 and for mating with edges 30 and 32 of lower container 12. Stria 46 run vertically along internal surface 44 from rib 40 to the internal surface of shoulder 48. Although not shown in the FIGS., stria 46 also run horizontally and radially along the internal surface of shoulder 48 toward the longitudinal axis of upper container 14. When upper container 14



is disposed on lower container 12, stria 34 of edges 30 and 32 couple with stria 46 to prevent upper container 14 and lower container 12 from rotating relative to each other.

As shown best in FIGS. 5-6, a plurality of sawteeth 50 are disposed on the external surface of shoulder 48. As shown in FIG. 7, sawteeth 50 mate with flexible wings 52 to

5 removably couple safety seal 19 to upper container 14. Safety seal 19 has a ring-shaped geometry with a perforation at connecting point 58. Safety seal 19 has an axial wing 56 that is connected to the remainder of seal 19 at a connecting point 59. Upper container 14 also has a neck 55 having external threads 56 for mating with internal threads 60 of cap 18, as is best shown in FIG. 8. As is best shown in FIG. 5, upper container 14 has bottom  
10 or non-permeable membrane 62 that is tearable along a perforated line 64 disposed about the periphery of bottom 62.

Tubular sleeve 16 includes a hollow body 64 having a truncated cone-shaped portion 66 on one end and a helicoidal edge 68 on an opposite end. Truncated cone-shaped portion 66 has a internal channel 70 terminating in a reverse truncated cone-shaped hole 72 that serves as the nozzle or dropper to dispense the mixed composition  
15 from device 10. Alternatively truncated cone-shaped portion 66 may be modified to include a conventional "Luer Lok" that complies with Luer Taper Specification 70.1 of the American Standards Association and that allows coupling to a syringe, cannula, or other conventional medical instruments. Helicoidal edge 68 preferably has a small  
20 horizontal section 74. Tubular sleeve 16 also includes a first sealing ring 76 and a second sealing ring 78 disposed on the external surface of body 64. Sealing rings 76 and 78 mate with internal surface 84 of reservoir 20 of upper container 14 to prevent leakage of first part 22 from reservoir 20 or entry of air into reservoir 20. Sealing ring 78 is especially useful in preventing such leakage or entry when first part 22 is a liquid. As shown in FIG.

8, cap 18 preferably has a member 80 that seals hole 72 when cap 18 is screwed onto upper container 14.

The above-referenced description is a summary of the structure of device 10. Certain portions of device 10 are described in greater detail in U.S. Patent Nos. 5,474,209  
5 and 5,782,345, which are incorporated herein by reference.

Referring generally to FIGS. 1-10, the preferred use of device 10 to mix the first part 22 and second part 26 of the second preferred composition of the present invention and to administer the mixed composition to the external ear canal will be described in greater detail. As shown in FIGS. 3-4, device 10 is in a first position in which the first  
10 part 22 and the second part 26 are unmixed. To mix parts 22 and 26, a user first moves axial wing 56 of safety seal 19 radially outward, breaking connecting point 59. The user then holds axial wing 56 and rotates safety seal 19 about a longitudinal axis of device 10, causing sawteeth 50 of upper container 14 to engage flexible wings 52 of safety seal 19. Such rotation splits safety seal 19 at connection point 58. The user can then remove  
15 safety seal 19 from device 10.

Next, the user screws cap 18 downward onto neck 55 of upper container 14. During this downward travel of cap 18, internal shoulder 85 of cap 18 contacts external shoulder 86 of tubular sleeve 16. Tubular sleeve is pushed downward within upper container 14 until external shoulder 86 of tubular sleeve 16 contacts shoulder 88 of upper  
20 container 14. Referring to FIG. 10, as tubular sleeve 16 is pushed downward, helicoidal edge 68 tears bottom 62 of reservoir 20 along perforation 64, with the exception of a portion of perforation 64 at horizontal section 74 of edge 68. Bottom 62 is thus opened but remains connected to upper container 14. First part 22 is then in communication with second part 26, and may be further mixed by shaking device 10, if necessary. In this  
25 second position of device 10 in which parts 22 and 26 are mixed, sealing rings 76 and 78

mate with internal surface 84 of upper container 14 to prevent leakage of the first or second parts of the composition from reservoirs 20 and 24 or entry of air into the reservoirs. Upon removal of cap 18, the mixed composition may be dispensed into the external ear canal by squeezing on the external surface of lower container 12 so as to force the mixture through nozzle 66 of tubular sleeve 16. When cap 18 is re-threaded onto tubular sleeve 16, member 80 seals hole 72.

A third preferred composition includes bicarbonate, a cerumenolytically acceptable enzyme, and an otologically acceptable vehicle. Sodium bicarbonate is the preferred bicarbonate. Sodium bicarbonate is preferably present in the amount of about 0.5% to about 15%. The amount of sodium bicarbonate is most preferably about 5%. The cerumenolytically acceptable enzyme for the third preferred composition is preferably identical to those described above in connection with the second preferred composition. The cerumenolytically acceptable enzyme is preferably present in the amount of about 50 AU/ml to about 1000 AU/ml. The most preferred cerumenolytically acceptable enzymes are proteolytic enzymes. The proteolytic enzyme is preferably present in the amount of about 50 AU/ml to about 1000 AU/ml. The most preferred proteolytic enzyme is methyl trypsin. Methyl trypsin is preferably present in the amount of about 50 AU/ml to about 500 AU/ml. Methyl trypsin is most preferably present in the amount of 200 AU/ml. The otologically acceptable vehicle is preferably identical, or substantially similar, to the otologically acceptable vehicle described above in connection with the second preferred composition. Examples 1, 2, 4, 5, and 7 hereinbelow provide in vitro data showing that the third preferred composition is efficacious in assisting in the removal of human and artificial cerumen.

Similar to the second preferred composition, the third preferred composition is preferably prepared and packaged as two separate parts, which are mixed prior to

administration to the external ear canal. The first part preferably comprises the cerumenolytically acceptable enzyme and the enzyme stabilizer of the otologically acceptable vehicle, and the second part preferably comprises the sodium bicarbonate and the remainder of the otologically acceptable vehicle. Examples 6, 8, 9, and 10

5       hereinbelow describe examples of the first part and the second part of the third preferred composition of the present invention, as well as a preferred method of making said first and second parts.

          Once made, the first and second parts may be removed from their separate packages and mixed. A bottle with a syringe or dropper, or a squeezable plastic dispenser  
10       bottle, is preferably used to deliver the mixed composition to ear as described hereinabove in connection with the first preferred composition. Alternatively, the third preferred composition may be packaged in and administered to the external ear canal via a single device having separate containers or compartments for the first part and the second part. A preferred device is device 10 described above in connection with the second  
15       preferred composition.

          As described hereinabove, all the compositions of the present invention may be packaged in a bottle with a syringe or dropper to dispense the composition, a plastic bottle that may be squeezed to dispense the composition, or a device such as device 10. For such compositions that also include a bicarbonate, it is preferred that the bottled  
20       compositions also be packaged in a blister pouch or pack made from aluminum foil in order to minimize any unwanted rise in the pH of the composition. When so packaged, such compositions, unlike currently known aqueous sodium bicarbonate based cerumenolytics, are pH stable at up to 40 degrees Celsius for up to 3 months after preparation.

In all of the compositions of the present invention that are packaged in a plastic bottle or device and that use benzalkonium chloride as a preservative, the otologically acceptable vehicle preferably also includes sodium citrate.2H<sub>2</sub>O or another citrate. Such bottles and devices are preferably made from a soft plastic, such as polyethylene, and are preferably sterilized via conventional ethylene oxide (ETO) methods. It has been unexpectedly discovered that citrate helps to maintain the concentration of benzalkonium chloride over time and thus yields a product that is more commercially viable.

All of the compositions of the present invention that comprise a liquid enzyme are preferably prepared in a sterile manner by filtration through an appropriate microbial filter. After the first and second parts of the compositions have been sterilized by filtration, a further advantage of the device 10 is that the first and second parts of the composition are admixed aseptically. This is because device 10 containing the first and second parts of the compositions is assembled in an aseptic manner with an air-tight seal, thus sterile mixing is performed within the closed sterile system of the device.

The following examples are intended to illustrate, but in no way limit, the present invention.

### **EXAMPLE 1**

Table 1 shows examples of the preferred compositions of the present invention, the compositions of certain components of these exemplary compositions, and the vehicles for these exemplary compositions. All ingredient amounts are in units of weight/volume percent unless otherwise indicated. Composition A is an example of the first preferred composition, composition B is an example of the second preferred composition after the mixing of its first and second parts, and compositions C and D are examples of the third preferred composition after the mixing of their first and second

parts, respectively. Composition A is also the “bicarbonate component of Composition C”, and Composition B is also the “proteolytic enzyme component of Composition C”. Composition D<sub>1</sub> is the “bicarbonate component of Composition D”, and Composition D<sub>2</sub> is the “proteolytic enzyme component of Composition D”. V<sub>1</sub> is the otologically acceptable vehicle for Compositions A, B, and C, and V<sub>2</sub> is the otologically acceptable vehicle for Composition D. Sodium chloride was added to V<sub>2</sub> and D<sub>2</sub> so that these compositions would have an equivalent salt content as V<sub>1</sub>.

5

TABLE I

Ingredient	A	B	C	D	V <sub>1</sub>	V <sub>2</sub>	D <sub>1</sub>	D <sub>2</sub>
Sodium Bicarbonate	5	---	5	5	---	---	5	---
Sodium Citrate. 2H <sub>2</sub> O	3	3	3	---	3	---	---	---
Sodium Chloride	---	---	---	---	---	3.48	---	3.48
Methyl Trypsin	---	200 AU/ml	200 AU/ml	200 AU/ml	---	---	---	200 AU/ml
Glycerin	7	7	7	7	7	7	7	7
Tetronic® 1304	0.25	0.25	0.25	---	0.25	---	---	---
Benzalkonium Chloride	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
pH	8.1	8.3	8.3	8.0	8.3	6.0	7.9	7.0
Water	q.s.	q.s.	q.s.	q.s.	q.s.	q.s.	q.s.	q.s.

## EXAMPLE 2

Compositions A and B were tested for their efficacy in assisting in the removal of cerumen in the following manner.

Samples of human cerumen were provided by various otology clinics. Samples were variable in type and amount. Therefore, about 20 to 30 specimens were pooled and mixed in a pestle and mortar to form a larger, relatively homogeneous sample lot. This enabled a series of assays to be carried out with a single lot of human cerumen.

In order to overcome the lot-to-lot variation in human cerumen, an artificial cerumen was developed. The artificial cerumen consists of a mixture of three components. The first component (50%) is a lipid mixture based upon the reported composition of the lipids in ear wax. The second component (30%) is homogenized bovine corneal epithelial cells, which simulates the desquamated epidermal cells in ear wax. The third component (20%) is lyophilized fetal bovine serum, which simulates the other components of ear wax that are secreted from the ceruminous and sebaceous glands.

### Preparation of 14 grams of Artificial Cerumen

1. The following lipids are weighed out into a glass vessel, and then dissolved in 70 mls of chloroform/methanol (2:1 v/v) with warming:



<b>Lipid Ingredient</b>	<b>Vendor</b>	<b>Amount (grams)</b>
Squalene	ICN Biochemicals, Inc. of Auora, Ohio ("ICN")	0.448
Cholesterol	Research Chemical Ltd., Heysham, Lancaster, England	1.463
Cholesterol Palmitate	ICN	0.336
Cholesterol Stearate	Sigma Chemical Co.	0.336
Oleic Acid OleylEster	Sigma Chemical Co.	0.326
Oleic Acid Stearyl Ester	Sigma Chemical Co.	0.326
Triolein	ICN	0.105
Oleic Acid	Mallinckrodt Baker, Inc. of Paris, Kentucky	2.097
Stearic Acid	Calbiochem Biosciences, Inc. of LaJolla, California	0.795
Cholesterol Sulfate	Sigma Chemical Co.	0.140
Phosphatidylcholine, dioleoyl	Sigma Chemical Co.	0.131
Phosphatidylcholine, distearoyl	Sigma Chemical Co.	0.131
Phosphatidylcholine, dipalmitoyl	Sigma Chemical Co.	0.131
Phosphatidylcholine, dimyristoyl	Sigma Chemical Co.	0.131
<b>Total</b>		<b>6.896</b>
Triglyceride concentrate Chicken egg yolk (suspension) 43.39mg/ml	Calbiochem Biosciences, Inc. of LaJolla, California	0.242 mls (0.105 grams)*
<u><b>Total</b></u>		<b>7.001</b>

\* = The triglyceride suspension is mixed with water in step 4.

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2. The bovine corneal epithelial cells were prepared as follows. Frozen bovine mature eyes (total of 305) were purchased from Pel-Freeze®. After thawing in water, the corneal epithelial cells were scraped from the eyes using a razor blade, and accumulated in 200 mls of deionized water. This suspension was treated with a Polytron®

homogenizer using a setting of 4 for 1 minute. The resulting suspension was then lyophilized to give 4.2 grams of dried bovine corneal epithelial cells.

3. The dried fetal bovine serum was prepared as follows. Fetal bovine serum (100 ml) was purchased from Hyclone and lyophilized to give 5.34 grams of a pale pink powder. A portion of this powder (2.8 grams) was used in the preparation of artificial cerumen.

4. The artificial cerumen was prepared as follows. Using a 5 inch glass mortar and pestle, the corneal epithelial cells (4.2 grams) and dried fetal bovine serum (2.8 grams) were mixed with a few drops of water to form a paste. To this paste was added the chloroform/methanol solution of the lipids (fume hood) and the mixture was ground with the mortar to produce a uniform paste. This was allowed to slowly evaporate in the fume hood until the odor of chloroform had disappeared. The artificial wax was then collected and stored in a sealed and air-tight container in a refrigerator (4°C). Before use in the efficacy testing, the artificial cerumen was warmed to 37°C.

#### **In Vitro Efficacy Testing (Human and Artificial Cerumen)**

About 30mg samples of the pooled lot of human cerumen or the artificial cerumen were accurately weighed using an analytical balance. Each sample was then gently rolled into a ball using the fingers (gloved), and placed in 12x75 mm borosilicate glass test tubes. Each test solution was pre-warmed and then added (2ml) to a cerumen sample in a water-bath set at 37°C, where it was allowed to stand without agitation.

At the time interval (30 minutes or 2 hours), the contents of each test tube were pipetted into separate syringes pre-fitted with Acrodisc® filters. Using finger pressure,

the sample was filtered into size 13 (100mm) borosilicate glass test-tubes. Some samples are difficult to filter completely, and so filtration attempts are stopped after 90 seconds, and the analysis then proceeds with the volume collected. This procedure is repeated in triplicate for each test solution. Staggering the digestions for 5 or 10 minutes can conveniently achieve this.

Absorbencies at 600nm were recorded usually without any dilution of the filtrate. In the event that the absorbency was above 0.7 units, or that there was less than 1 ml of filtrate, appropriate dilutions were made. Absorbencies at 280nm were made on diluted solutions of the filtrate. A suitable dilution is usually 0.2 ml of the filtrate combined with 4 ml of water (factor of 21). The absorbency of each original test solution was also recorded at both wavelengths using the same dilutions as the corresponding test sample.

The cerumenolytic activity of the test solution is defined as the absorbance units released by the test solution from a gram of cerumen per ml of solution, after the incubation. For calculations, the absorbance reading of the solution blank with an identical dilution to the test sample solution was subtracted from the sample reading, and this net absorbance was then multiplied by the dilution factor and the 2 ml test volume to obtain the total units. The total absorbance units were then divided by the weight, in grams, of the cerumen sample, as follows:

$$((\text{Absorbency of test solution} - \text{absorbency of solution blank}) \times 2 \times \text{dilution factor}) / \text{sample weight in grams} = \text{absorbency/gram/ml}$$

The absorbency at 280nm indicates the amount of protein digested by the composition, while the absorbency at 600nm indicates the amount of color, lipids and other ingredients released from the cerumen.

## Results

Table 2 represents the amount of cerumen digested by each composition as measured by a spectrophotometer in absorbency units per gram of cerumen per ml of composition after the cerumen has been immersed in the composition for 2 hours. The 280 nanometer wavelength is referred to herein as the “protein component”. The 600 nanometer wavelength is referred to herein as the “lipid component”. As shown by Table 2, Compositions A and B were each effective in digesting the protein component and the lipid component of both the human and artificial cerumen.

**TABLE 2**

Efficacy Results  
(Absorbency Units Per Gram Per ml at 2 Hours)

Cerumen	A	B	C	V <sub>1</sub>	A - V <sub>1</sub>	B - V <sub>1</sub>	(A - V <sub>1</sub> ) + (B - V <sub>1</sub> )	C - V <sub>1</sub>
Human 280 nm	1659	1654	1740	1120	+ 539	+ 534	+ 1073	+ 620
Human 600 nm	8.0	10.6	9	11.2	- 3.2	- 0.6	- 3.8	- 2.2
Artificial 280 nm	365	166	854	166	+ 199	0	+ 199	+ 688
Artificial 600 nm	18.8	6.4	52	4.6	+ 14.2	+ 1.8	+ 16	+ 47.4

## EXAMPLE 3

Table 3 shows the composition of the first part of Composition B before mixing with the second part of Composition B. Table 4 shows the composition of the second part of Composition B before mixing with the first part of Composition B. Composition B is preferably made by simple mixing of the composition of Table 3 with the composition of Table 4. All ingredient amounts are in units of weight/volume percent unless otherwise indicated.

**TABLE 3**  
**(First Part of Composition B)**

INGREDIENT	AMOUNT
Glycerin	46.7
Methyl Trypsin	1335 AU/ml
Purified water	q.s.

5

**TABLE 4**  
**(Second Part of Composition B)**

INGREDIENT	AMOUNT
Tetronic®1304	0.294
Benzalkonium Chloride	0.012
Sodium Citrate.2H <sub>2</sub> O	3.529
Citric Acid	q.s. to pH 8.0
Purified water	q.s.

10

Various volumes of the first part and the second part of Composition B may be employed. The preferred ratio of the volume of the first part of Composition B to the volume of the second part of Composition B is 3:17. A preferred volume of the first part is 1.5 ml. A preferred volume of the second part is 8.5 ml. The glycerin in the first part of Composition B acts as an enzyme stabilizing agent for the methyl trypsin. After the first part and the second part of Composition B are mixed, the glycerin also acts as a demulcent.

15

#### **EXAMPLE 4**

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Composition C was tested for its efficacy in assisting in the removal of human cerumen in the manner described in Example 2. Table 2 represents the amount of cerumen digested by Composition C as measured by a spectrophotometer in absorbency

units per gram of cerumen per ml of composition after the cerumen as been immersed in the composition for 2 hours. As shown by Table 2, Composition C was effective in digesting the protein component and the lipid component of both the human and artificial cerumen.

In addition, it was unexpectedly discovered that composition C exhibits a synergistic effect in digesting artificial cerumen. Referring again to Table 2, the absorbency measurements for the bicarbonate component A of Composition C, the proteolytic enzyme component B of Composition C, and the vehicle V<sub>1</sub> of Composition C are also provided. As shown by Table 2, the sum of the absorbency measurements of the bicarbonate component A and the proteolytic enzyme component B, subtracting any contribution from vehicle V<sub>1</sub>, is significantly less than the absorbency measurement for Composition C, subtracting any contribution from vehicle V<sub>1</sub>, for both the protein component and the lipid component of artificial cerumen.

### EXAMPLE 5

Composition C was also tested for its efficacy in removing cerumen as compared to Murine® Ear Drops and Cerumenex® Eardrops. Composition C, Murine, and Cerumenex were tested in the manner described in Example 2, except that absorbency values were measured at 30 minutes instead of 2 hours.

Table 5 shows the amount of cerumen digested by Composition C, Cerumenex, and Murine as measured by a spectrophotometer in absorbency units per gram of cerumen per ml of composition after the cerumen has been immersed in the composition for 30 minutes. As shown by Table 5, Composition C was much more effective in removing the protein component of both human and artificial cerumen than either Murine or Cerumenex. Composition C was also much more effective in removing the lipid

component of artificial cerumen than either Murine or Cerumenex. In addition, as shown by Table 5 and Table 8 (discussed hereinbelow), Composition C was much more effective in removing the protein and lipid components of artificial cerumen than a five percent sodium bicarbonate solution.

**TABLE 5**

Efficacy Results  
(Absorbency Units Per Gram Per ml at 30 Minutes)

Cerumen	C	Murine <sup>®</sup>	Cerumenex <sup>®</sup>
Human 280 nm	970	53	133
Human 600 nm	8	14	9
Artificial 280 nm	628	27	65
Artificial 600 nm	85	0	3

**EXAMPLE 6**

Table 6 shows the composition of the first part of Composition C before mixing with the second part of Composition C. Table 7 shows the composition of the second part of Composition C before mixing with the first part of Composition C. Composition C is preferably made by simple mixing of the composition of Table 6 with the composition of Table 7. All ingredient amounts are in units of weight/volume percent unless otherwise indicated.

**TABLE 6**  
**(First Part of Composition C)**

INGREDIENT	AMOUNT
Glycerin	46.7
Methyl Trypsin	1335 AU/ml
Purified water	q.s.

**TABLE 7**  
**(Second Part of Composition C)**

INGREDIENT	AMOUNT
Tetronic® 1304	0.29
Sodium Bicarbonate	5.9
Benzalkonium Chloride	0.012
Sodium Citrate.2H <sub>2</sub> O	3.53
Citric Acid	q.s. pH 8.0
Purified water	q.s.

5

Various volumes of the first part and the second part of Composition C may be employed. The preferred ratio of the volume of the first part of Composition C to the volume of the second part of Composition C is 3:17. A preferred volume of the first part is 1.5 ml. A preferred volume of the second part is 8.5 ml. The glycerin in the first part of Composition C acts as an enzyme stabilizing agent for the methyl trypsin. After the first part and the second part of Composition C are mixed, the glycerin also acts as a demulcent.

10

**EXAMPLE 7**

Composition D was tested for its efficacy in assisting in the removal of cerumen as compared to an aqueous solution of five percent sodium bicarbonate. Composition D and the five percent sodium bicarbonate solution were tested in the manner described in Example 2, except that absorbency values were measured at 30 minutes instead of 2 hours.

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Table 8 shows the amount of cerumen digested by Composition D and the five percent sodium bicarbonate solution as measured by a spectrophotometer in absorbency



units per gram of cerumen per ml of composition after the cerumen has been immersed in the composition for 30 minutes. As shown by Table 8, Composition D was much more effective in removing the lipid component of human cerumen and the protein and lipid components of artificial cerumen than the five percent sodium bicarbonate solution.

5           In addition, it was unexpectedly discovered that composition D exhibits a synergistic effect in removing both human and artificial cerumen. Referring again to Table 8, the absorbency measurements for the bicarbonate component  $D_1$  of Composition D, the proteolytic enzyme component  $D_2$  of Composition D, and the vehicle  $V_2$  of Composition D are also provided. As shown by Table 4, the sum of the absorbency  
10       measurements of the bicarbonate component  $D_1$  and the proteolytic enzyme component  $D_2$ , subtracting any contribution from vehicle  $V_2$ , is significantly less than the absorbency measurement for Composition D, subtracting any contribution from vehicle  $V_2$ , for the protein component and the lipid component of both human and artificial cerumen.

**TABLE 8**  
**Efficacy Results**  
**(Absorbency Units Per Gram Per ml at 30 Minutes)**

Cerumen	D <sub>1</sub>	D <sub>2</sub>	D	5% Sodium Bicarbonate Solution	V <sub>2</sub>	D <sub>1</sub> -V <sub>2</sub>	D <sub>2</sub> -V <sub>2</sub>	(D <sub>1</sub> -V <sub>2</sub> ) + (D <sub>2</sub> -V <sub>2</sub> )	D-V <sub>2</sub>
Human 280 nm	897	716	1153	1169	675	+ 222	+41	+263	+478
Human 600 nm	12.2	2.9	14.0	8.8	3.0	+9.2	-0.1	+9.1	+11.0
Artificial 280 nm	557	49	812	577	31	+526	+18	+544	+781
Artificial 280 nm	40	1.2	49	39	0.6	+39.4	+0.6	+40.0	+48.4

**EXAMPLE 8**

Table 9 shows the composition of the first part of Composition D before mixing with the second part of Composition D. Table 10 shows the composition of the second part of Composition D before mixing with the first part of Composition D. Composition D is preferably made by simple mixing of the composition of Table 9 with the composition of Table 10. All ingredient amounts are in units of weight/volume percent unless otherwise indicated.

**TABLE 9**  
**(First Part of Composition D)**

INGREDIENT	AMOUNT
Glycerin	46.7
Methyl Trypsin	1335 AU/ml
Purified water	q.s.

**TABLE 10**  
**(Second Part of Composition D)**

INGREDIENT	AMOUNT
Sodium Bicarbonate	5.88
Benzalkonium Chloride	0.012
Hydrochloric Acid	q.s. to pH 8.0
Purified water	q.s.

Various volumes of the first part and the second part of Composition D may be employed. The preferred ratio of the volume of the first part of Composition D to the volume of the second part of Composition D is 3:17. A preferred volume of the first part is 1.5 ml. A preferred volume of the second part is 8.5 ml. The glycerin in the first part of Composition D acts as an enzyme stabilizing agent for the methyl trypsin. After the

first part and the second part of Composition D are mixed, the glycerin also acts as a demulcent.

### **EXAMPLE 9**

5      Example 9 is another example of the third preferred composition of present invention containing calcium chloride and boric acid as enzyme stabilizing agents. Table 11 shows the first part of the composition, which is preferably housed in upper container 14 of device 10. Table 12 shows the second part of the composition, which is preferably housed in lower container 12 of device 10. Table 13 shows the mixed composition,  
10      which is preferably made by simple mixing of the composition of Table 11 with the composition of Table 12 in the preferred volume ratio of 3:17. All ingredient amounts are in units of weight/volume percent unless otherwise indicated. The glycerin in the first part of the composition acts as an enzyme stabilizing agent for the methyl trypsin. After the first part and the second part of the composition are mixed, the glycerin also acts as a  
15      demulcent.

**TABLE 11**  
**(First Part of Composition)**

<b>INGREDIENT</b>	<b>AMOUNT</b>
Glycerin	46.7
Methyl Trypsin	1335 AU/ml
Boric Acid	1.5
Calcium Chloride	0.25
NaOH/HCl	Adjust to pH 7
Purified water	q.s.

5

**TABLE 12**  
**(Second Part of Composition)**

<b>INGREDIENT</b>	<b>AMOUNT</b>
Tetronic®1304	0.294
Sodium Bicarbonate	5.882
Benzalkonium Chloride	0.012
Sodium Citrate.2H <sub>2</sub> O	3.529
Citric Acid	q.s. to pH 8.0
Purified water	q.s.

10

**TABLE 13**  
**(Mixed Composition)**

INGREDIENT	AMOUNT
Methyl Trypsin	200 AU/ml
Glycerin	7.0
Sodium Bicarbonate	5.0
Tetronic®1304	0.25
Boric Acid	0.225
Calcium Chloride	0.0375
Benzalkonium Chloride	0.01
Sodium Citrate.2H <sub>2</sub> O	3.0
Citric acid	q.s. to pH 8.0
Purified water	q.s.

5

**EXAMPLE 10**

Table 14 shows another example of the composition of a first part of Composition C before mixing with the second part of Composition C. Table 15 shows another example of the composition of the second part of Composition C before mixing with the first part of Composition C. All ingredient amounts are in units of weight/volume percent unless otherwise indicated.

10

**TABLE 14**  
**(First Part of Composition C)**

INGREDIENT	AMOUNT
Methyl Trypsin	2.22 mg (2000 AU)
Sodium Citrate.2H <sub>2</sub> O	187.4 mg
Citric Acid (Anhydrous)	4.2 mg
Glycerin	2.0 mg
Tetronic® 1304	4.0 mg
Dehydrated Alcohol	28 mg

**TABLE 15**  
**(Second Part of Composition C)**

INGREDIENT	AMOUNT
Benzalkonium Chloride	0.01
Glycerin	6.98
Sodium Bicarbonate	5.00
Sodium Citrate.2H <sub>2</sub> O	1.13
Tetronic® 1304	0.21
Citric Acid	q.s. pH 8.0
Purified water	q.s. to 100 ml

5

The first part of composition C is a powder or granulate. The composition of Example 10 is thus preferred when it is desired for the first part of the composition to be in powder or granulate form. Various masses and volumes of the first part and the second part of Composition C may be employed, respectively. The preferred mass of the first part is 200 mg, and the quantities in Table 14 are for 200 mg of the first part of Composition C. The preferred volume of the second part is 10 ml. 1 mg of methyl trypsin equals 900 AU. The dehydrated alcohol component of the first part is evaporated during manufacture. When the first part and second parts are mixed, the resulting Composition C has an identical formulation to Composition C shown in Table 1 of Example 1. The glycerin in the first part of Composition C acts as an enzyme stabilizing agent for the powder or granulate methyl trypsin, and as a binder between the sodium citrate.2H<sub>2</sub>O and anhydrous citric acid. The glycerin in the second part of Composition C, and in the composition after the first part and the second part of Composition C are mixed, also acts as a demulcent. The pH of the first part of Composition C is preferably about 6.5.

20

Composition C of Example 10 is preferably made using the following technique. The first part of Composition C, as shown in Table 14, is preferably prepared as follows. 187.4 mg of sodium citrate.2H<sub>2</sub>O is passed through a high-speed, rotary, 450  $\mu$ m sieve into a planetary mixer. 2.0 mg of glycerin is then dissolved under agitation in 14 mg of dehydrated alcohol in a compounding vessel. The compounding vessel is preferably made of glass or stainless steel. The alcohol is preferably anhydrous ethanol. 4.2 mg of anhydrous citric acid is then added to the compounding vessel and allowed to dissolve under agitation. The contents of the compounding vessel are then added to the sodium citrate.2H<sub>2</sub>O in the planetary mixer, and the resulting mixture is mixed in the planetary mixer for 10 minutes. The mixture will form a granulate. The granulate fixed in the shovels of the planetary mixer is then removed, and the mixture is mixed in the planetary mixer for an additional 5 minutes. The wet granulate is removed from the planetary mixer, extended onto a tray, covered with a cloth, and allowed to dry at room temperature for at least 12 hours. The dried granulate is passed through a high speed, rotary, 450  $\mu$ m sieve to form a preliminary granulate.

Methyl trypsin is milled using a Waring blender that is preferably rotating at 18,000 to 25,000 rpm, and most preferably at 21,000 rpm. Approximately 20 percent excess methyl trypsin (2.7 mg total) is used to compensate for losses during manufacture. Small quantities of the preliminary granulate and small quantities of the methyl trypsin are alternatively added to a clean planetary mixer and mixed until homogeneous. This mixing typically takes about 10 minutes. 4.0 mg of Tetronic® 1304 is then dissolved under agitation in 14 mg of dehydrated alcohol in a compounding vessel. The compounding vessel is preferably made of glass or stainless steel. The alcohol is preferably anhydrous ethanol. The Tetronic® 1304 and anhydrous ethanol are then added to the homogeneous mixture of preliminary granulate and methyl trypsin in the planetary



mixer and are mixed for 5 minutes. The Tetronic acts as a binder for the methyl trypsin and the preliminary granulate. The granulate fixed in the shovels of the planetary mixer is then removed, and the mixture is mixed in the planetary mixer for an additional 5 minutes. The wet granulate is removed from the planetary mixer, extended onto a tray, covered with a cloth, and allowed to dry at room temperature for at least 12 hours. The dried granulate is passed through a high-speed, rotary, 450  $\mu\text{m}$  sieve to form the final granulate or powder of the first part of Composition C. It has been discovered that the final granulate or powder of the first part of Composition C, when prepared using the above-described technique, has a stable concentration of methyl trypsin at up to 40 degrees Celsius for at least 3 months after preparation.

The second part of Composition C, as shown in Table 15, is preferably prepared as follows. The amounts in parenthesis represent the amounts for a 10 ml volume of the second part of Composition C. An amount of purified water equivalent to about 50% of the required volume (5.0 ml) is transferred to a suitable compounding vessel. 0.21 weight/volume percent (.021 g) of Tetronic® 1304 is then added to the compounding vessel and allowed to dissolve. The Tetronic® 1304 in the second part of Composition C, and in the composition after the first part and the second part of Composition C are mixed, also acts as a surfactant. Next, 5.00 weight/volume percent (0.5 g) of sodium bicarbonate is added to the compounding vessel under agitation at a low speed and allowed to dissolve. 6.98 weight/volume percent (0.698 g) of glycerin is then added to the compounding vessel under agitation and allowed to dissolve. Next, 0.01 weight/volume percent plus an additional 5% excess to compensate for loss during manufacture (0.0021 ml of a 50 weight/volume percent solution in water) of benzalkonium chloride is added to the compounding vessel under agitation and allowed to mix. 1.13 weight/volume percent sodium citrate.2H<sub>2</sub>O (0.113 g) is then added to the compounding vessel under agitation

and allowed to dissolve. The pH of the mixture in the compounding vessel is then adjusted to 8.0 by adding citric acid. The citric acid is preferably anhydrous. Purified water is then added to the compounding vessel under agitation to reach the desired volume of the second part of Composition C of 10 ml. The final pH of the mixture is then checked. Next, the mixture is filtered through a 0.22  $\mu$ m membrane to obtain the final second part of Composition C. Composition C is then made by simple mixing of the composition of Table 14 with the composition of Table 15.

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From the above, it may be appreciated that the present invention provides improved compositions for assisting in the removal of human cerumen. In addition, the compositions of the present invention are useful for the treatment of other otic and nasal conditions. For example, given the presence of a tympanostomy tube or other rupture or hole in the tympanic membrane, the compositions of the present invention may be used to cleanse the middle and external ear of the viscous exudate that often results from a middle ear infection. This viscous exudate, which is commonly called “glue ear”, includes conditions such as secretory otitis media, mucoid otitis media, serous otitis media, and chronic otitis media with effusion. These conditions sometimes lead to hearing loss via inflammation and/or the inhibition of ossicular sound conduction by the exudate. As another example, the compositions of the present invention may be applied to the inner tissues of the nose via irrigation or spray to dissolve nasal crusts or scabs that result from surgery or other insult or trauma to the nasal tissue. Such crusts and scabs are typically a mixture of blood, mucus, and secretions from open mucosal surfaces. As a further example, the compositions of the present invention may be used to dissolve other crusts or scabs.

The present invention has been described by reference to certain preferred embodiments. However, it should be understood that it may be embodied in other specific forms or variations thereof without departing from its spirit or essential characteristics. The embodiments described above are therefore considered to be

5 illustrative in all respects and not restrictive, the scope of the invention being indicated by the appended claims rather than by the foregoing description.